

Exercise Sheet 1 due 16 April 2018

1. atomic units

Express

- the speed of light
- the Bohr magneton

in atomic units.

2. magnetic moment

From classical magnetostatics we know that the magnetic moment due to an electrical current density \vec{j}_e is given by

$$\vec{m} = \frac{1}{2} \int \vec{r} \times \vec{j}_e d^3r .$$

- i. Given the quantum-mechanical probability current density

$$\vec{j} = \frac{\hbar}{2im_e} \left(\overline{\psi(\vec{r})} \vec{\nabla} \psi(\vec{r}) - \psi(\vec{r}) \vec{\nabla} \overline{\psi(\vec{r})} \right) ,$$

calculate the corresponding magnetic moment. Compare to the expectation value of the angular momentum operator \vec{L} .

- ii. What is the z-component of the magnetic moment for the following orbitals of the hydrogen atom $|n, l, m\rangle$: $|1, 0, 0\rangle$, $|2, 0, 0\rangle$, $|2, 1, -1\rangle$, $|2, 1, 0\rangle$, $|2, 1, 1\rangle$, and $|5, 3, 2\rangle$. Express your results using the Bohr-magneton

$$\mu_B = \frac{e\hbar}{2m_e} .$$

3. charge states

What formal charge do you expect for the transition metal in KCrF_3 ? Which for the manganese ions in SrMnO_3 ?

4. atomic radii

Read the article W.L. Bragg, *Phil. Mag.* **40**, 169 (1920) and try to understand how atomic radii are derived from crystal structure data.

Historical notes: First crystal structures were determined 1912 by Max von Laue (Nobel Prize in Physics 1914 *for his discovery of the diffraction of X-rays by crystals*) and by the Braggs (father and son, Nobel Prize 1915 *for their services in the analysis of crystal structure by means of X-rays*). The Schrödinger equation was published 1926 (Nobel Prize 1933). For a readable account of the discoveries, you can browse the Nobel lectures at http://www.nobelprize.org/nobel_prizes/physics/laureates/1914 etc..